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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.004 Å R factor = 0.049 wR factor = 0.132 Data-to-parameter ratio = 13.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the title compound, $C_9H_7ClN_2O_2$, the occurrence of $O-H \cdots N$ hydrogen-bond interactions results in the formation of a pseudo-dimer arranged around an inversion center.

[3-(2-Chlorophenyl)-1,2,4-oxadiazol-5-yl]methanol

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Comment

1,2,4-Oxadiazoles represent an important class of fivemembered heterocycles. Some derivatives of 1,2,4-oxadiazoles have anti-inflammatory (Nicolaides *et al.*, 1998) and antipicornaviral (Romero, 2001) properties. We are focusing our synthetic and structural studies on new oxadiazole derivatives and we recently published the synthesis and structure of [3-(2methylphenyl)-1,2,4-oxadiazol-5-yl]methanol (Yan *et al.*, 2006). We report here the structure of its close analogue, (I), in which 2-methylphenyl is replaced by a 2-chloro group,.



The molecular structure of (I), shown in Fig. 1, is roughly planar, the dihedral angle between the benzene ring and the oxadiazole ring being only $6.2 (1)^{\circ}$. There are $O-H\cdots N$



Figure 1

The structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme. H atoms are shown as small spheres of arbitrary radii.

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organic papers

intermolecular hydrogen bonds, resulting in the formation of a pseudo-dimer arranged around an inversion center (Table 1 and Fig. 2).

Experimental

Hexamethylenetetramine (90 mmol) was dissolved in acetic acid (70 ml) and water (70 mmol). 3-(2-Chlorophenyl)-5-chloromethyl-1,2,4-oxadiazole (30 mmol) was added to this mixture. The resulting mixture was refluxed for 3 h. After cooling and filtering, crude compound (I) was obtained. It was crystallized from a mixture of ethyl acetate (6 ml) and petrolum ether (6 ml). Crystals of (I) suitable for X-ray diffraction were obstained by slow evaporation of an ethanol solution.

Z = 4

 $D_{\rm v} = 1.573 {\rm Mg m}^{-3}$

Mo $K\alpha$ radiation

Block, colourless

 $0.40 \times 0.30 \times 0.30$ mm

 $\mu = 0.40 \text{ mm}^{-1}$

T = 293 (2) K

Crystal data

 $\begin{array}{l} C_9H_7{\rm CIN}_2{\rm O}_2\\ M_r = 210.62\\ {\rm Monoclinic,}\ P2_1/n\\ a = 7.4280\ (15)\ {\rm \AA}\\ b = 14.202\ (3)\ {\rm \AA}\\ c = 8.8050\ (18)\ {\rm \AA}\\ \beta = 106.76\ (3)^\circ\\ V = 889.4\ (3)\ {\rm \AA}^3 \end{array}$

Data collection

Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\min} = 0.856, T_{\max} = 0.889$ 1874 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.049$ $wR(F^2) = 0.132$ S = 1.091739 reflections 128 parameters H-atom parameters constrained 1739 independent reflections 1399 reflections with $I > 2\sigma(I)$ $R_{int} = 0.017$ $\theta_{max} = 26.0^{\circ}$ 3 standard reflections every 200 reflections intensity decay: none

$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0683P)^{2} + 0.2802P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.27 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.44 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O1\!-\!H1\!\cdots\!N1^i$	0.82	2.04	2.849 (3)	170
Symmetry code: (i)	-x + 1, -y + 1,	-z + 1.		

All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C-H = 0.95



Figure 2

Partial packing view, showing the formation of a pseudo-dimer through intermolecular $O-H \cdots N$ hydrogen bonds (dashed lines). [Symmetry codes: (i) -x + 1, -y + 1, -z + 1.]

(aromatic) or 0.99 Å (methylene) and O–H = 0.82 Å, with $U_{iso}(H) = 1.2U_{eq}(C,O)$

Data collection: *CAD-4 Software* (Enraf–Nonius,1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996), *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2003)'; software used to prepare material for publication: *SHELXL97*.

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